

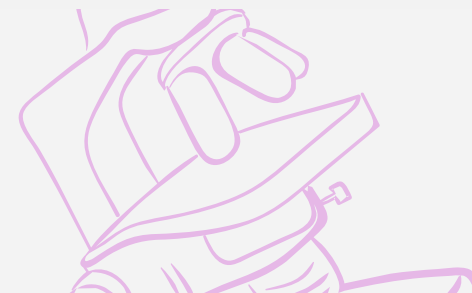
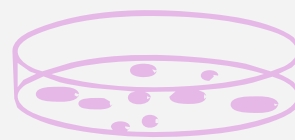
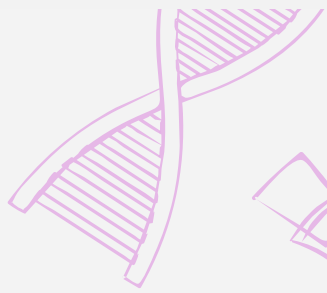
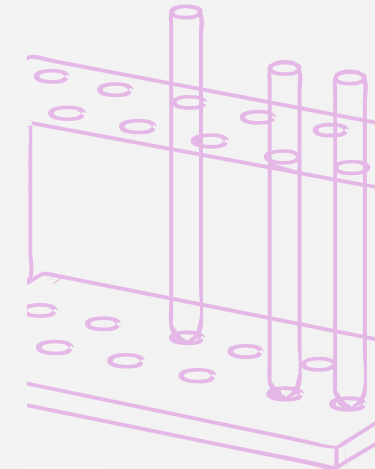
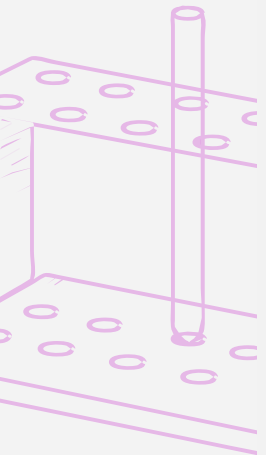
كلية الصيدلة
Faculty of Pharmacy

Pharmacophores

And Bioisosteres

Imbarkah Sulaiman (2659)

Salsabeel Alsherif (2376)



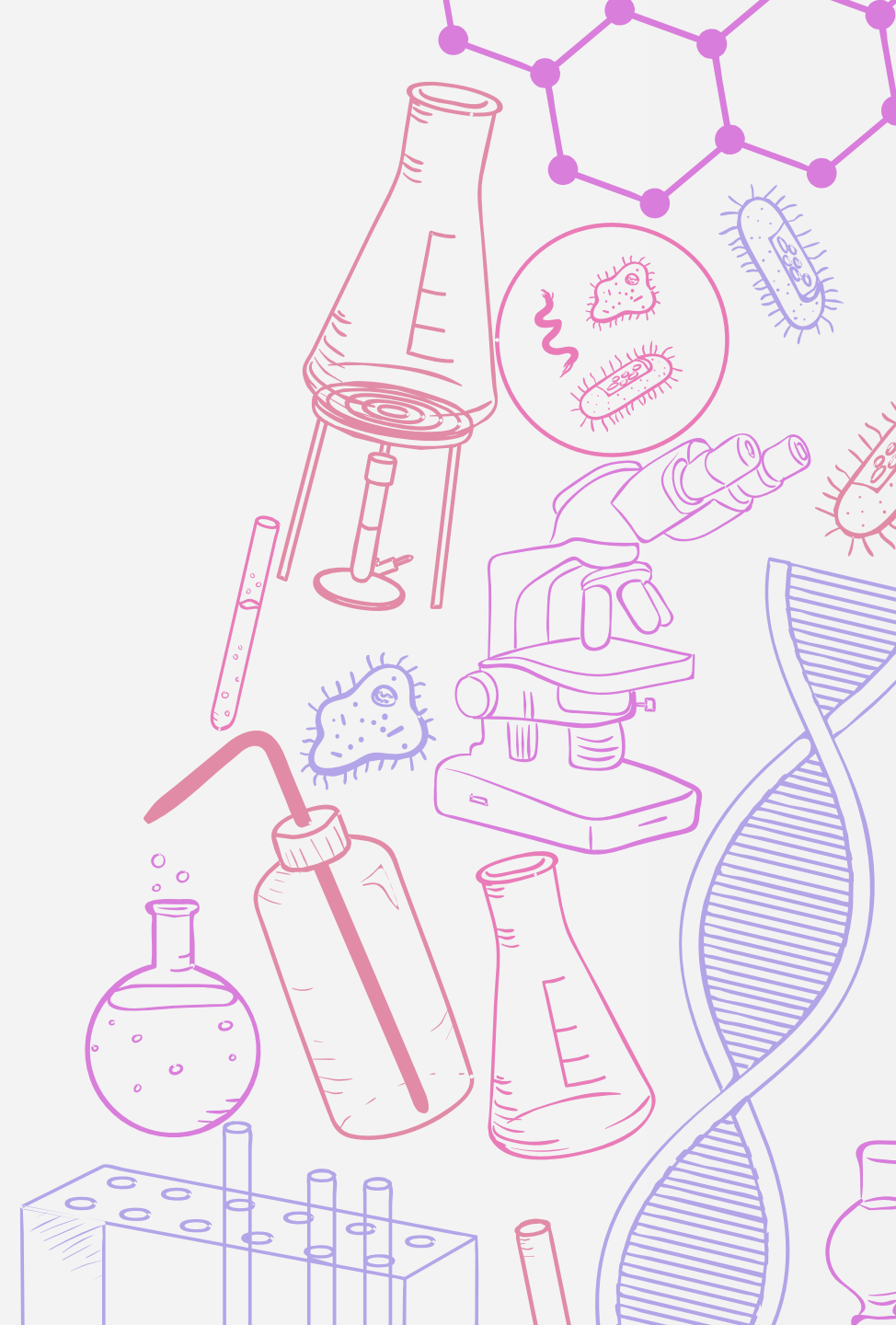


Intended Learning Outcomes

- **Define pharmacophores**
- **Classify Pharmacophores**
- **Mention applications of pharmacophores**
- **Define Bioisosteres**
- **Classify Bioisosteres**
- **Mention applications of Bioisosteres**

Pharmacophore

A **pharmacophore** is the ensemble of steric and electronic features that is necessary to ensure the optimal supramolecular interactions with a specific biological target structure and to trigger (or to block) its biological response.





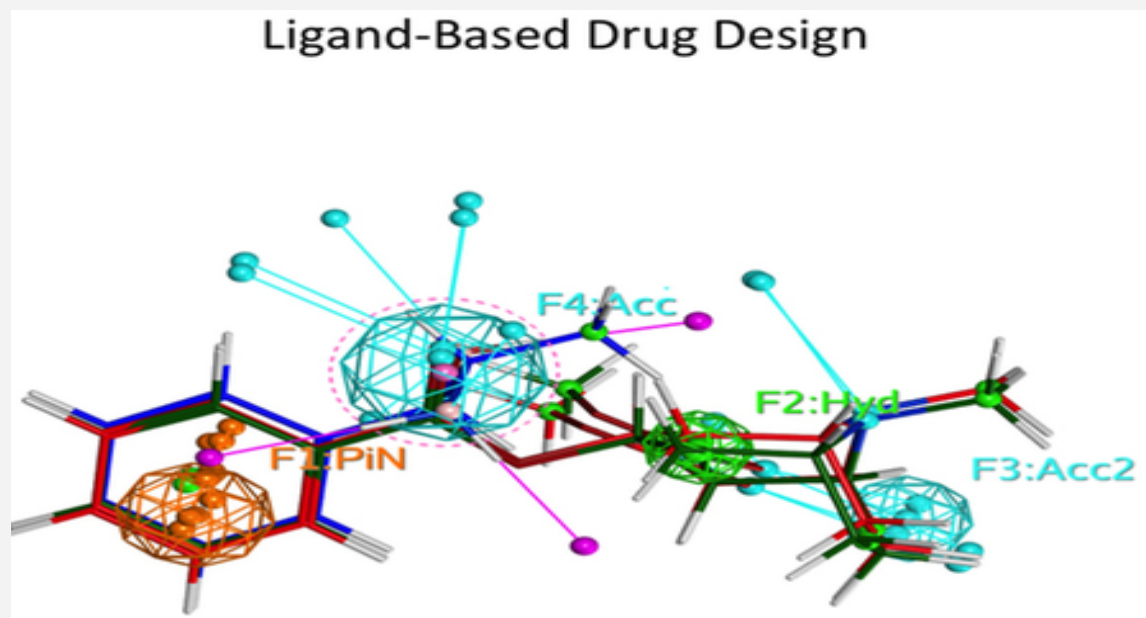
Pharmacophore Classification

Ligand based

Structure based

Ligand based

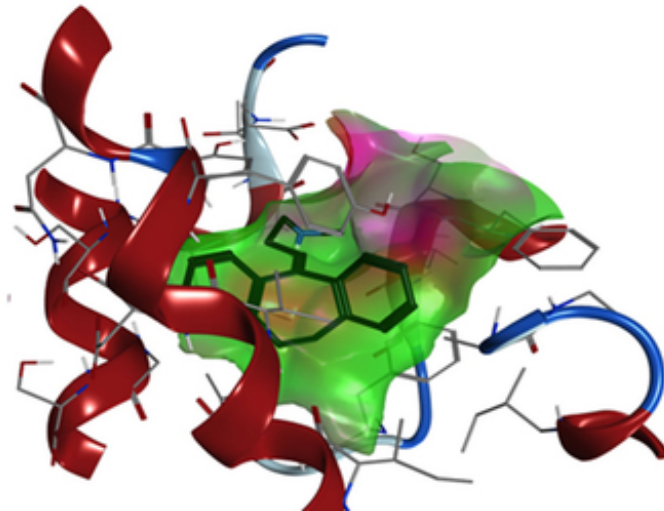
In the absence of the macromolecular target structure, ligand-based pharmacophore modeling is an essential strategy for drug discovery. In this method, the common chemical characteristics from 3D structures of multiple known ligands are extracted through ligand alignment, which would represent the essential interactions between ligand and potential macromolecular target.



Structure based

The structure - based pharmacophore modeling generates chemical features of the active site and the sterical relationships from 3D structure of macromolecular target of macromolecule - ligand complex. It probes the possible interaction sites between the macromolecular target and the ligands.

Structure-Based Drug Design



Pharmacophore Applications

Ligand based

**Virtual
Screening**

**Lead
Optimization**

Structure based

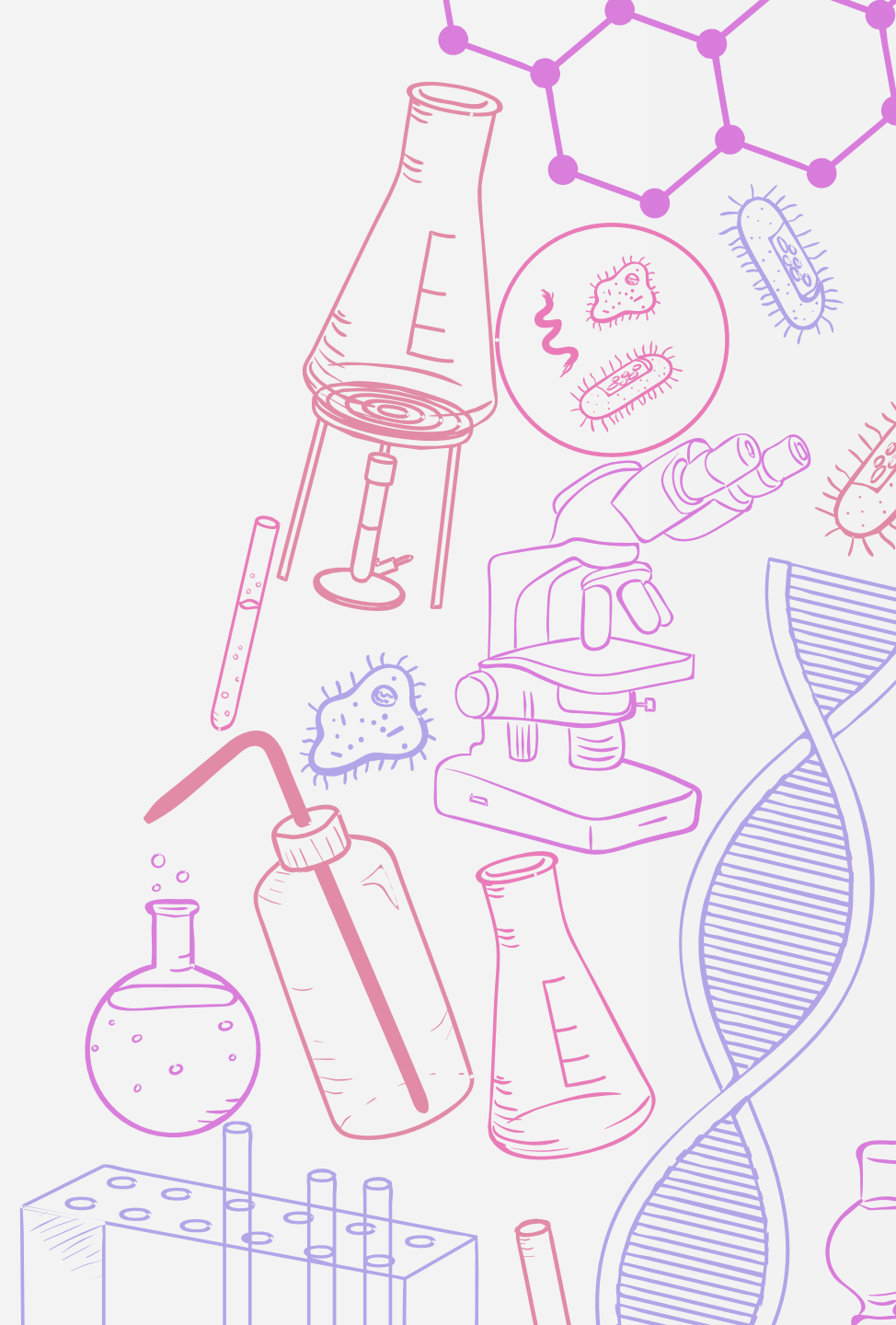
**De Novo
Drug Design**

**Target
Identification**



Bioisostere

Bioisosteres are chemical substituents or groups with similar physical or chemical properties which produce broadly similar biological properties to another chemical compound.





Bioisostere Classification



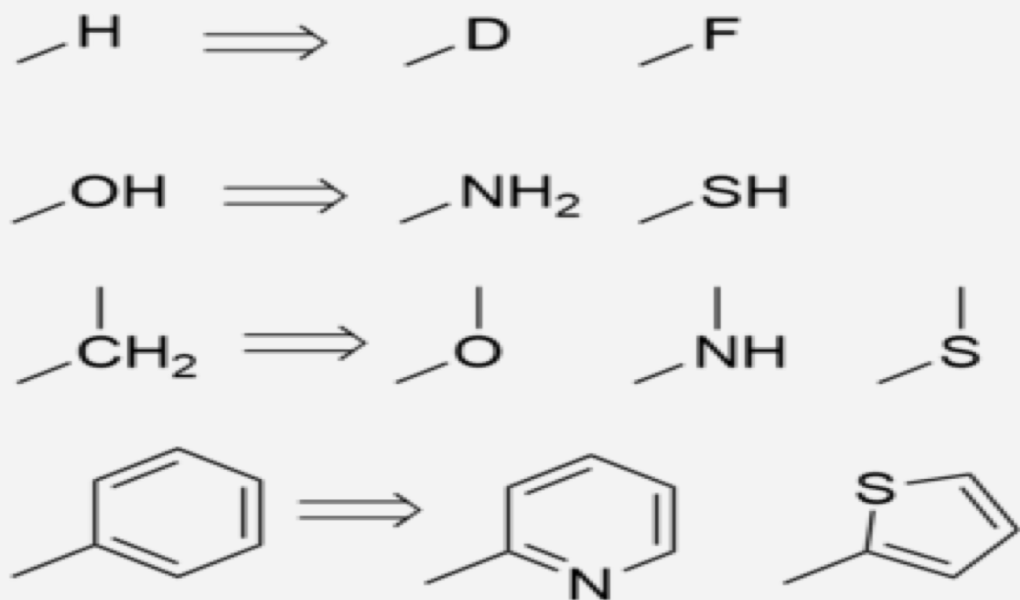
Classical bioisosteres

Non-Classical bioisosteres

Classical bioisosteres

Have same steric and electronic properties

1. Monovalent
2. Divalent
3. Trivalent
4. Tetravalent
5. Ring Equivalent

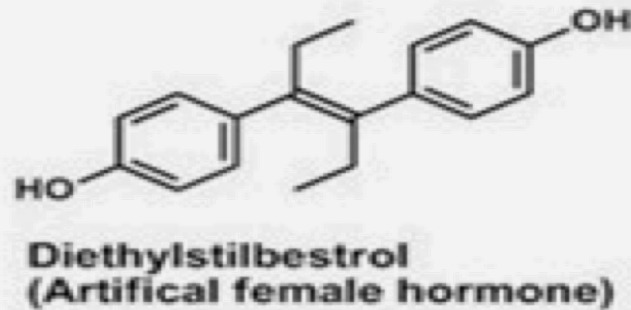
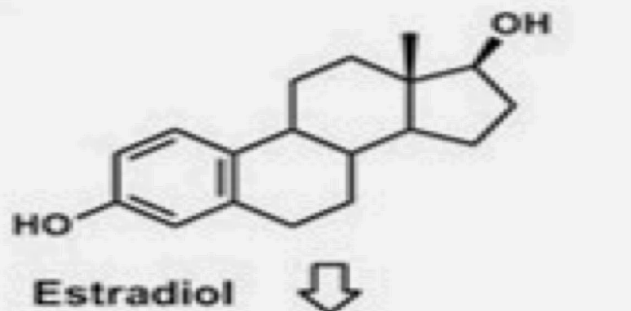


Non-Classical bioisosteres

Do not have steric and electronic properties

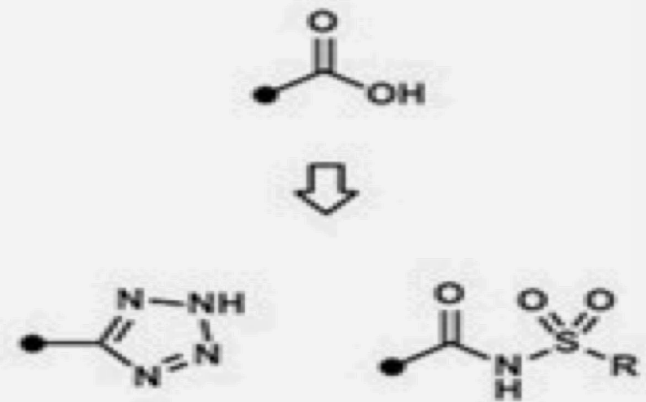
Non-classical bioisosteres divide to two subgroups,

○ Cyclic and non-cyclic isosteres



○ Exchangeable group

ex.) Carboxylic acid isosteres





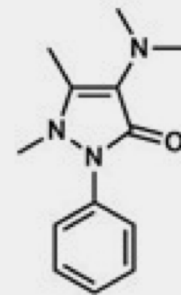
Summary:

The knowledge of a pharmacophore, or the 3D arrangement of features in the biologically active molecule that is responsible for its pharmacological activity, can help in the search and design of a new or better drug acting upon the same or related target.

A bioisostere is a molecule resulting from the exchange of an atom or of a group of atoms with an alternative, broadly similar, atom or group of atoms. The objective of a bioisosteric replacement is to create a new molecule with similar biological properties to the parent compound.

Bioisostere Application

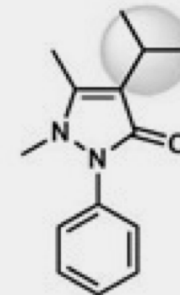
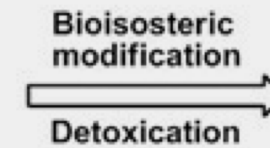
In drug design, the purpose of exchanging one bioisostere for another is to enhance the desired biological or physical properties of a lead compound without making significant changes in chemical structure of lead



Aminopyrine

Aminopyrine

Marketed as an analgesic and anti-inflammatory drug in 1896. In 1922, It was revealed that Aminopyrine was a carcinogen !



Propylphenazone

Propylphenazone

Developed by Roche in 1951. Bioisosteric modification of dimethylamino group removed its carcinogenic action.



**Thanks for
listening!**



References:

<https://slideplayer.com/slide/10412082/>

<https://www.slideshare.net/tubakhan10/bioisostersm>

<https://www.slideshare.net/abhikseal/pharmacohoreppt>

<https://www.future-science.com/doi/10.4155/fmc.11.18>