

# Pharmacophore and Bioisosteres

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# Objectives

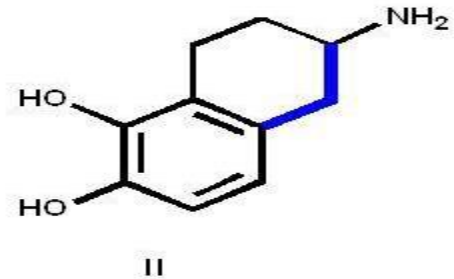
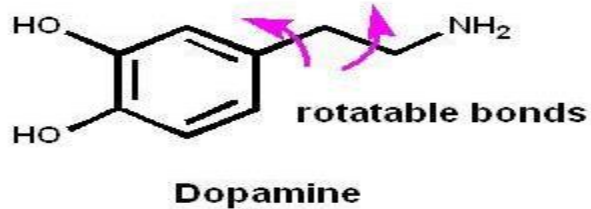
- 1 . Define pharmacophore.**
- 2 . List pharmacophore types of functional groups.**
- 3 . Describe pharmacophore modeling.**
- 4 . Define bioisosteres.**
- 5 . Classify bioisosteres.**
- 6 . Describe why bioisosteres are replaced.**



**Pharmacophore**

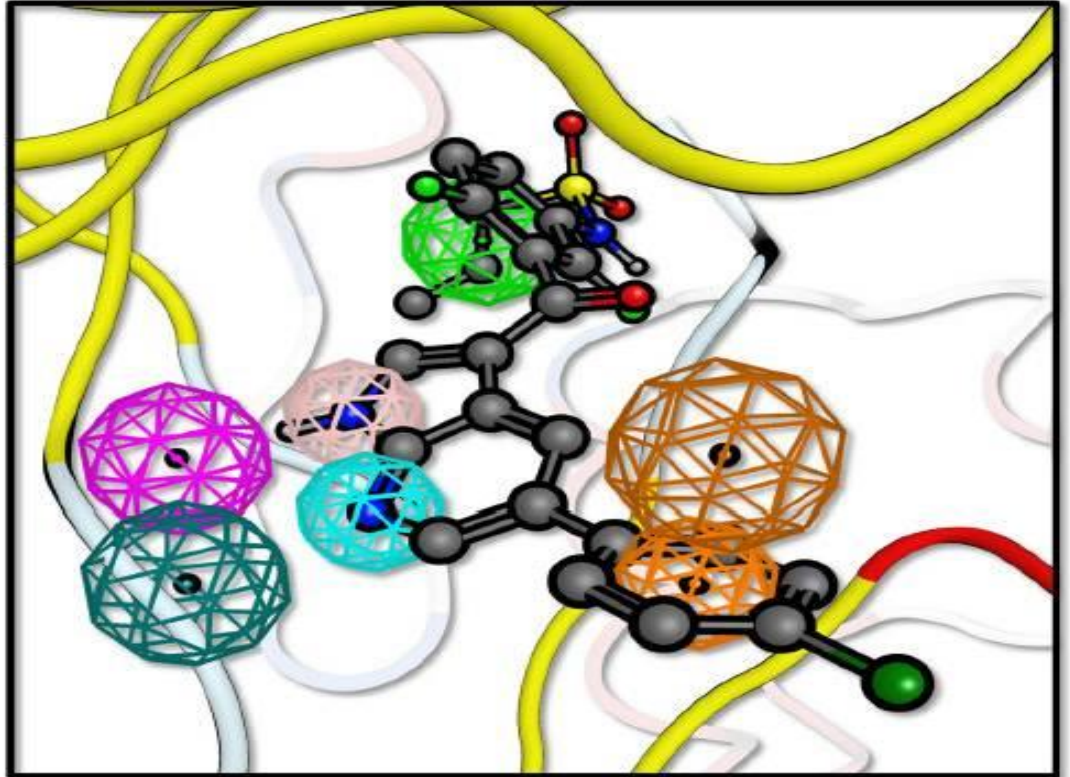
# Definition of Pharmacophore

Is the region of the molecule containing the essential organic functional groups that directly interact with the receptor active site and, therefore, confers the biologic activity of interest.



— Locked bonds

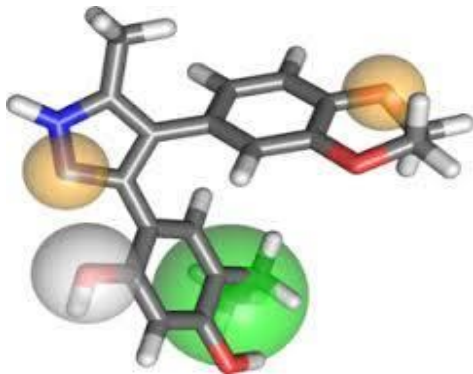
# Pharmacophore Types of Functional Groups



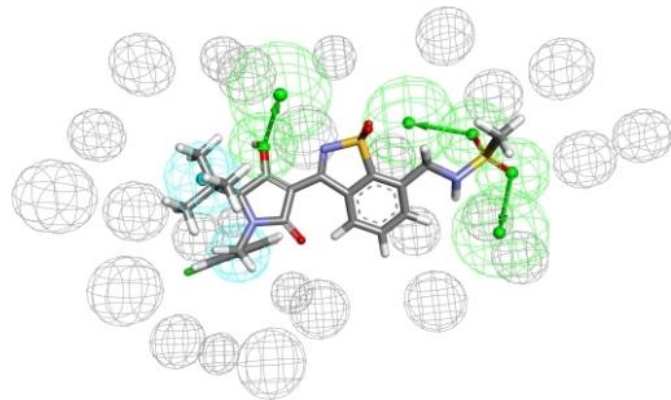
# Pharmacophore Modeling

Pharmacophore

Ligand Based



Structure Based



# Pharmacophore Molecule Feature

- **3D: (Hydrophobic group, charged ionizable group, H donor or acceptor).**
- **2D: (Substructures).**
- **1D: (Physical or biological properties).**

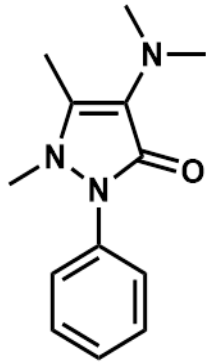


# Bioisosteres

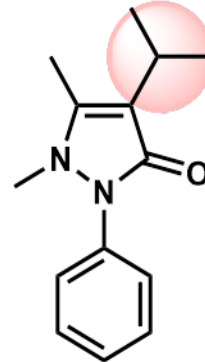
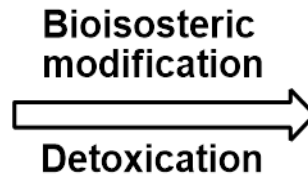


# Definition of Bioisosteres

Is the replacement or modification of specific pharmacophore (functional) groups with other groups having similar properties is known as **“isosteric replacement”** or **“bioisosteric replacement.”**



Aminopyrine



Propylphenazone

# Classification





# Classical Bioisosteres

**Classical bioisosteres represent the result of a nearly appreciation of the concept and encompass structurally simple atoms or groups.**

## Monovalent bioisosteres

F, H

OH, NH<sub>2</sub>

F, OH, NH<sub>2</sub> or CH<sub>3</sub> for H

SH, OH

Cl, Br, CF<sub>3</sub>

## Divalent bioisosteres:

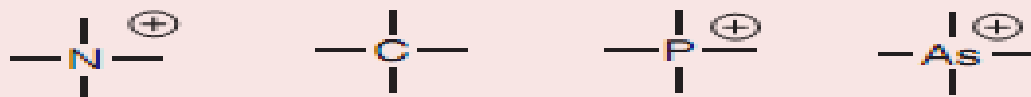
—C=S, —C=O, —C=NH, —C=C—

## Trivalent atoms or groups:

— $\begin{array}{c} \text{C} \\ | \\ \text{H} \end{array}$ —, —N=

—P=, —As=

## Tetrasubstituted atoms:



## Ring equivalents:

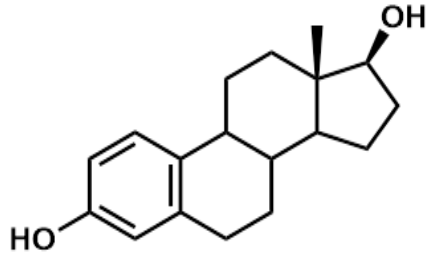




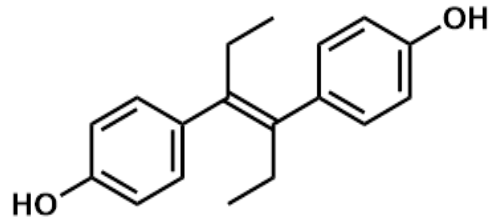
# **Non-Classical Bioisosteres**

**Non classical bioisosteres are structurally distinct, usually comprised of different number of atoms and exhibit different steric and electronic properties.**

## 1) Cyclic and noncyclic isostere



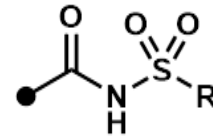
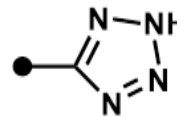
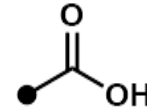
Estradiol



Diethylstilbestrol  
(Artificial female hormone)

## 2) Exchangeable group

ex.) Carboxylic acid isosteres



**Bioisosteres are replaced,  
Why ?!**





**Improved selectivity**

**Fewer side effect**

**Decreased toxicity**

**Improved  
pharmacokinetics**

**Increase stability**

**Patented lead  
compounds**







**All the bioisosteric not produce utilization  
biologic activity.**

- **True**

- **False**

# Summary

- ✓ **Pharmacophore** : it is the group of atoms in the molecule of a drug responsible for the drug's action.
- ✓ **Ligand-based and structure-based methods** have been developed for improving pharmacophore modeling .
- ✓ **bioisosterism”** to describe functional groups related in structure that have similar biologic effects.
- ✓ **There are two general types of bioisosteres: classical and non- classical.**
- ✓ **Bioisosterism is used to reduce toxicity, change bioavailability, or modify the activity of the lead compound, and may alter the metabolism of the lead.**

# References

- ❑ Chemistry, F. is principle of medicinal. (2000). Foye's principles of medicinal chemistry.
- ❑ <https://www.profacgen.com/pharmacophore-modeling.htm>.
- ❑ <https://www.slideshare.net/tubakhan10/bioisostersm>.
- ❑ [http://www.f.utokyo.ac.jp/~kanai/seminar/pdf/Lit\\_Y\\_Morita\\_M1.pdf](http://www.f.utokyo.ac.jp/~kanai/seminar/pdf/Lit_Y_Morita_M1.pdf).



**Thank you..**